

CCU: Consultative Committee for Units

President:

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Executive Secretary:

Dr Estefanía de Mirandés



Bureau

International des

Poids et

Mesures

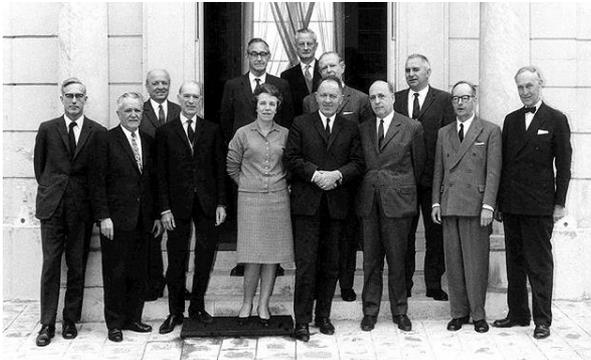
Date established:

1964, to replace the "Commission for the System of Units", set up by the CIPM in 1954

CCU: Consultative Committee for Units



23rd meeting of the CCU (2017)



1st meeting of the CCU (1967)

- provide advice about units of measurement
- develop the International System of Units (SI)
- prepare the *SI Brochure*

CCU: Consultative Committee for Units



23rd meeting of the CCU (2017)

Members of the CCU

- 11 National Metrology Institutes
- 3 Personal members
- 1 Ex officio member

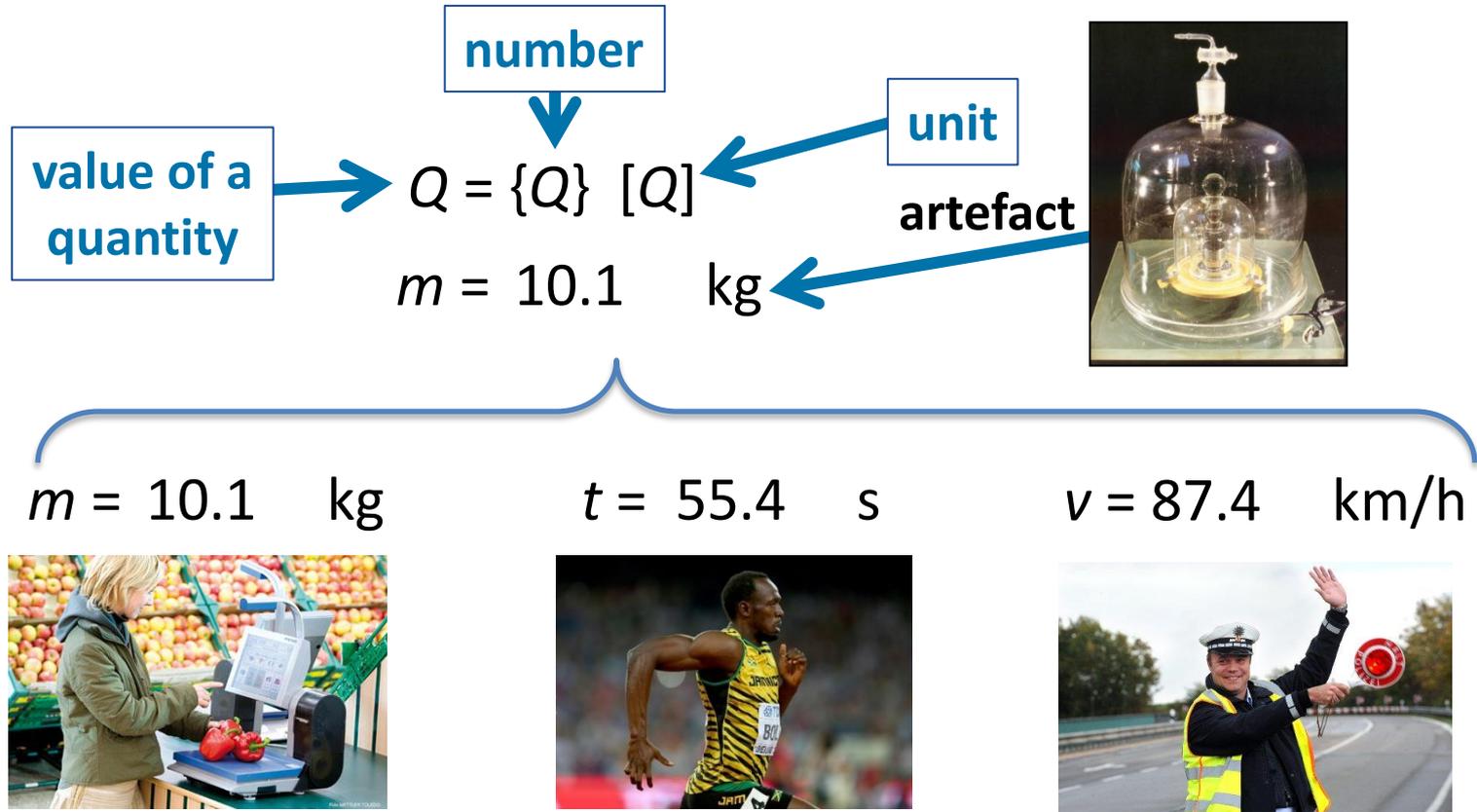
Liaisons of the CCU

- 3 International Unions
- 5 International Commissions & Committees
- 2 Intergovernmental Organizations or International bodies

CCU: Consultative Committee for Units

- ➔ Global forum for progressing the state-of-the-art
→ CCU is the only global forum for units
- ➔ Facilitating dialogue between NMIs and all stakeholders
→ Always through international consensus
- ➔ Global comparability of measurements

Quantities and Measurement Units



Quantities and Measurement Units

number

value of a quantity

$Q = \{Q\} [Q]$

unit

artefact

material property

measur. prescription

defining

uncertainty

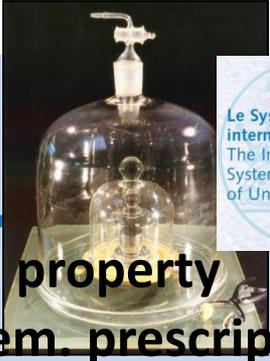
constant of nature

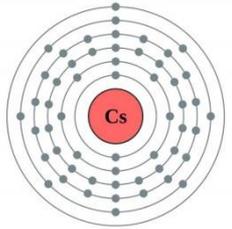
$m = 10.1(2) \text{ kg}$

$t = 55.4(1) \text{ s}$

$\Delta \nu(^{133}\text{Cs})_{\text{hfs}} = 9\,192\,631\,770.5(6) \text{ s}^{-1}$

$1 \text{ s} = \frac{9\,192\,631\,770}{\Delta \nu(^{133}\text{Cs})_{\text{hfs}}}$





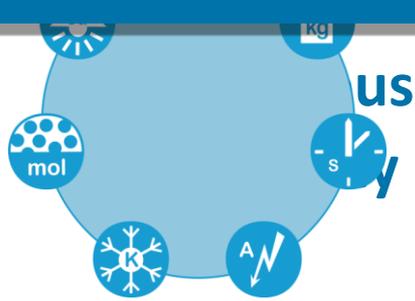


Define a unit by fixing the numerical value of a constant of nature

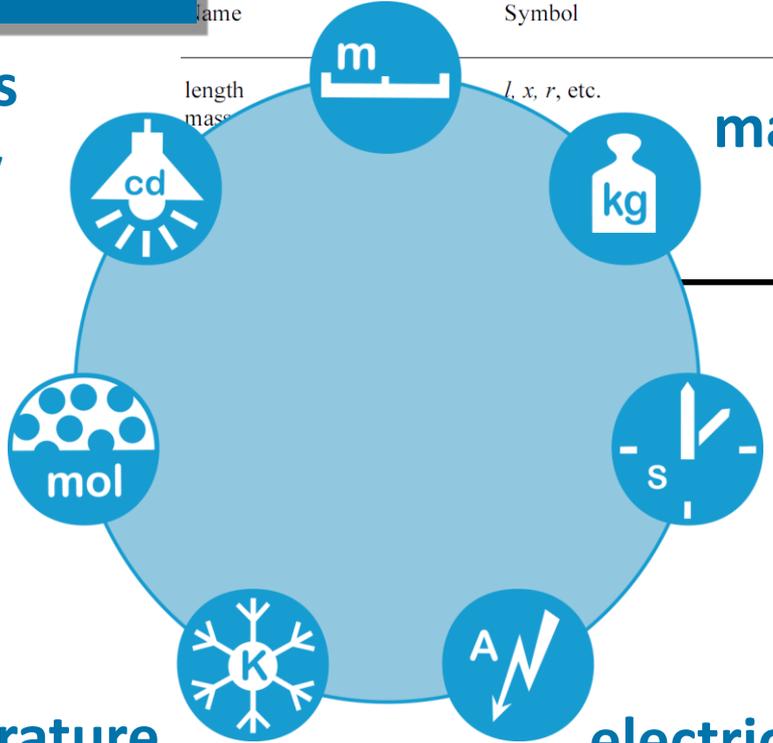
The International System of Units: SI

Established in 1960: CGPM

Base units



amount of substance



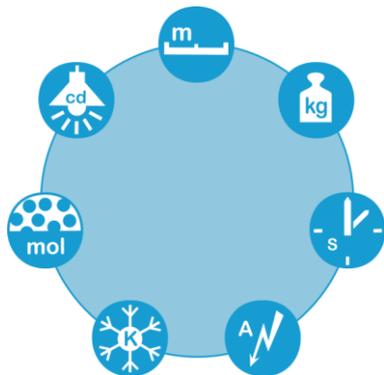
temperature

electric current

SI base unit	Name	Symbol
metre	metre	m
kilogram	kilogram	kg
second	second	s
ampere	ampere	A
kelvin	kelvin	K
mole	mole	mol
candela	candela	cd



The International System of Units: SI



Derived units

$$[v] = \text{m s}^{-1}$$

$$[c] = \text{mol m}^{-3}$$

Base units

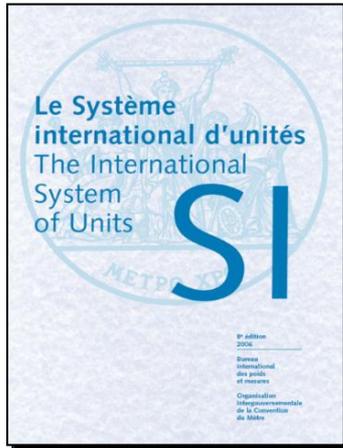
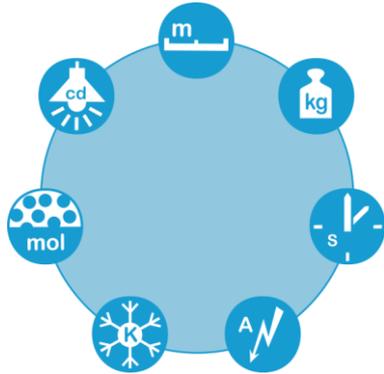
Derived units with special names

Dimensions of quantities

→ A set of coherent SI units

Quantity	SI base unit	Symbol	Expressed in terms of other SI units	Expressed in terms of SI base units
force	newton	N		m kg s^{-2}
pressure, stress	pascal	Pa	N/m^2	$\text{m}^{-1} \text{kg s}^{-2}$
energy, work, amount of heat	joule	J	N m	$\text{m}^2 \text{kg s}^{-2}$
power, radiant flux	watt	W	J/s	$\text{m}^2 \text{kg s}^{-3}$
electric charge, amount of electricity	coulomb	C		s A
electric potential difference, electromotive force	volt	V	W/A	$\text{m}^2 \text{kg s}^{-3} \text{A}^{-1}$
capacitance	farad	F	C/V	$\text{m}^{-2} \text{kg}^{-1} \text{s}^4 \text{A}^2$
electric resistance	ohm	Ω	V/A	$\text{m}^2 \text{kg s}^{-3} \text{A}^{-2}$
electric conductance	siemens	S	A/V	$\text{m}^{-2} \text{kg}^{-1} \text{s}^3 \text{A}^2$
frequency	hertz ^(d)	Hz		s^{-1}
force	newton	N		m kg s^{-2}

The International System of Units: SI



Base units

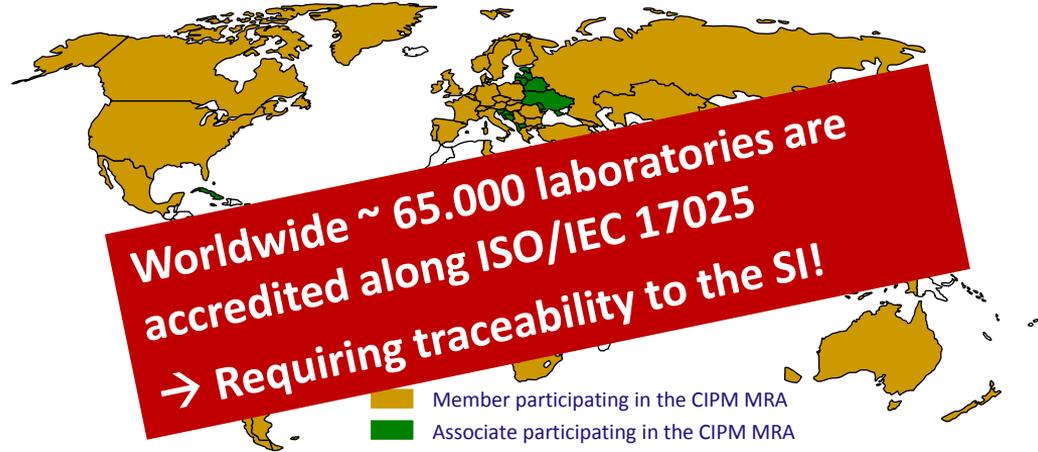
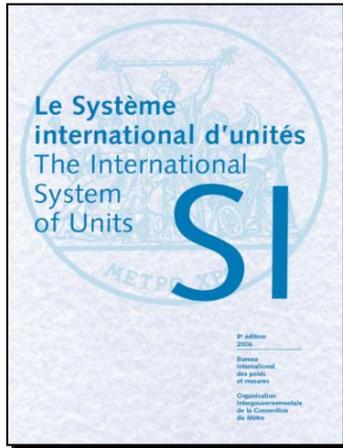
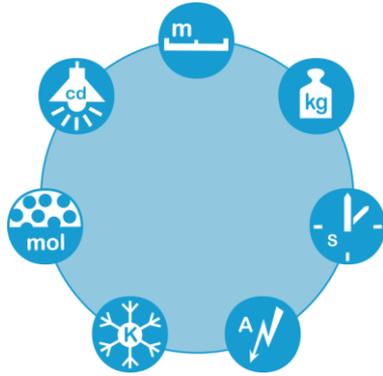
Derived units

Dimensions of quantities

→ A set of coherent SI units

- **A global measurement infrastructure**
- **Validation of light from an LED**
 - **CO₂ concentration in the air**
 - **Creatinine concentration in blood serum**
 - **Dose equivalent outside nuclear reactors**
 -

The International System of Units: SI



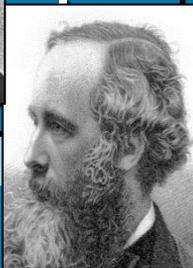
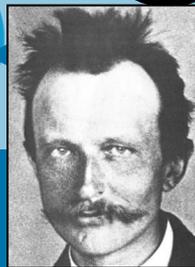
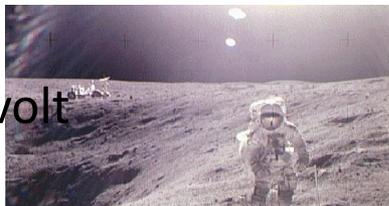
- A global measurement infrastructure
- Valid world wide: CIPM-MRA signed by
 - ➔ 97.6% of the world economy
 - ➔ The cornerstone of international quality infrastructure (QI)

The International System of Units: SI

Redefinitions

- 1967: second
- 1983: metre
- 1990: ohm, volt

$$1 \text{ m} = \frac{c}{299\,792\,458} \text{ s} \quad \text{length} \quad \boxed{c}$$



2018: $\sim 2 \times 10^{-16}$

time $\Delta \nu_{\text{hfs}}$

$$R_K = \frac{h}{e^2} \approx 25\,813 \, \Omega \Leftrightarrow R_{K-90}$$

$$K_J = \frac{2e}{h} \approx 483\,598 \text{ GHz/V} \Leftrightarrow K_{J-90}$$

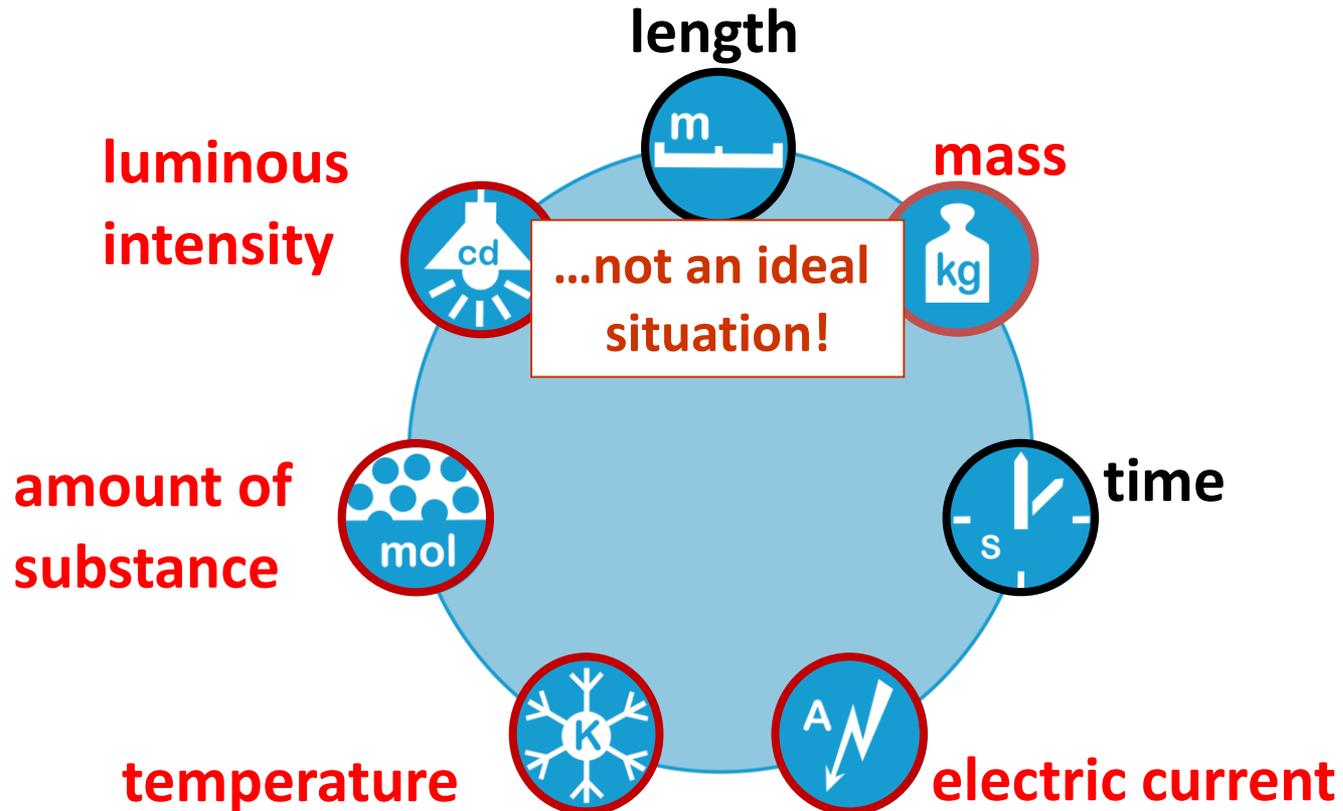
$$1 \text{ s} = \frac{9\,192\,631\,770}{\Delta \nu(^{133}\text{Cs})_{\text{hfs}}}$$

$$R_K, K_J \Delta \nu(^{133}\text{Cs})_{\text{hfs}}$$

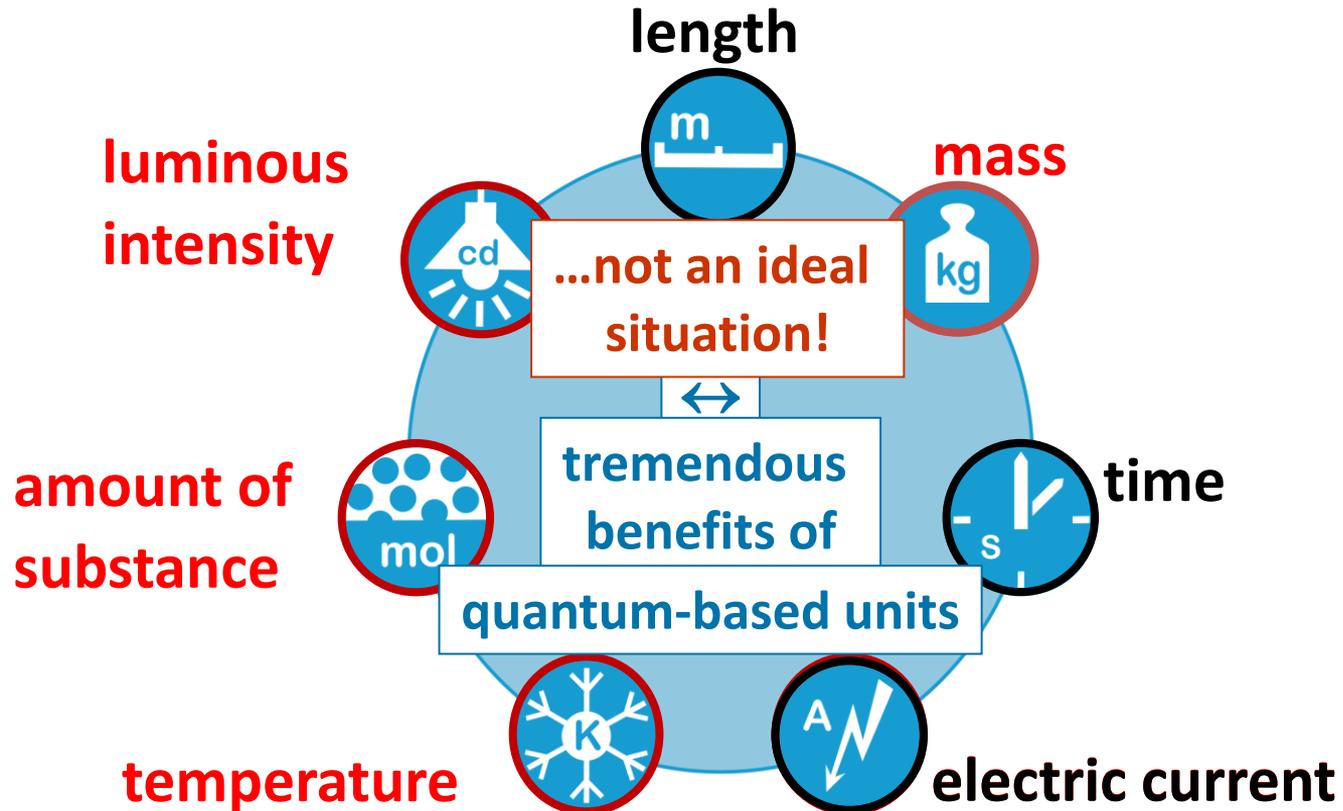


electric current

The International System of Units: SI

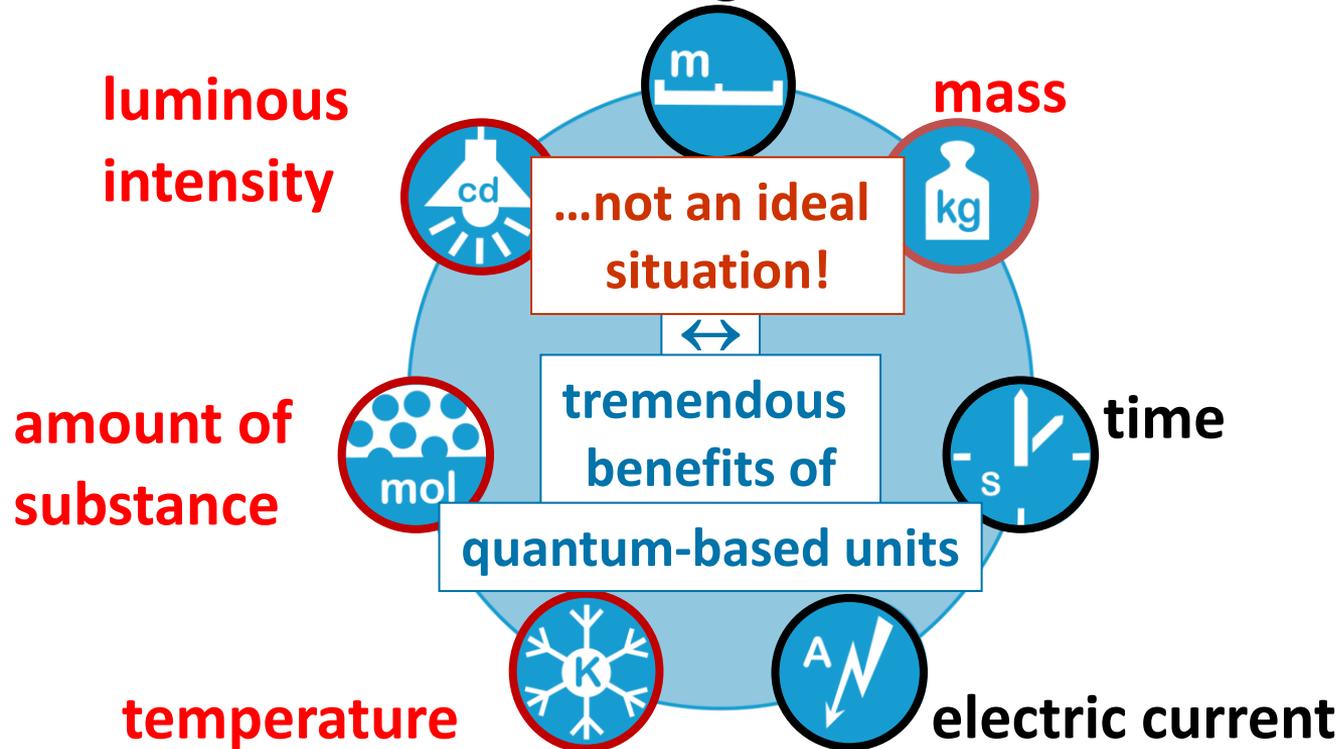


The International System of Units: SI



The International System of Units: SI

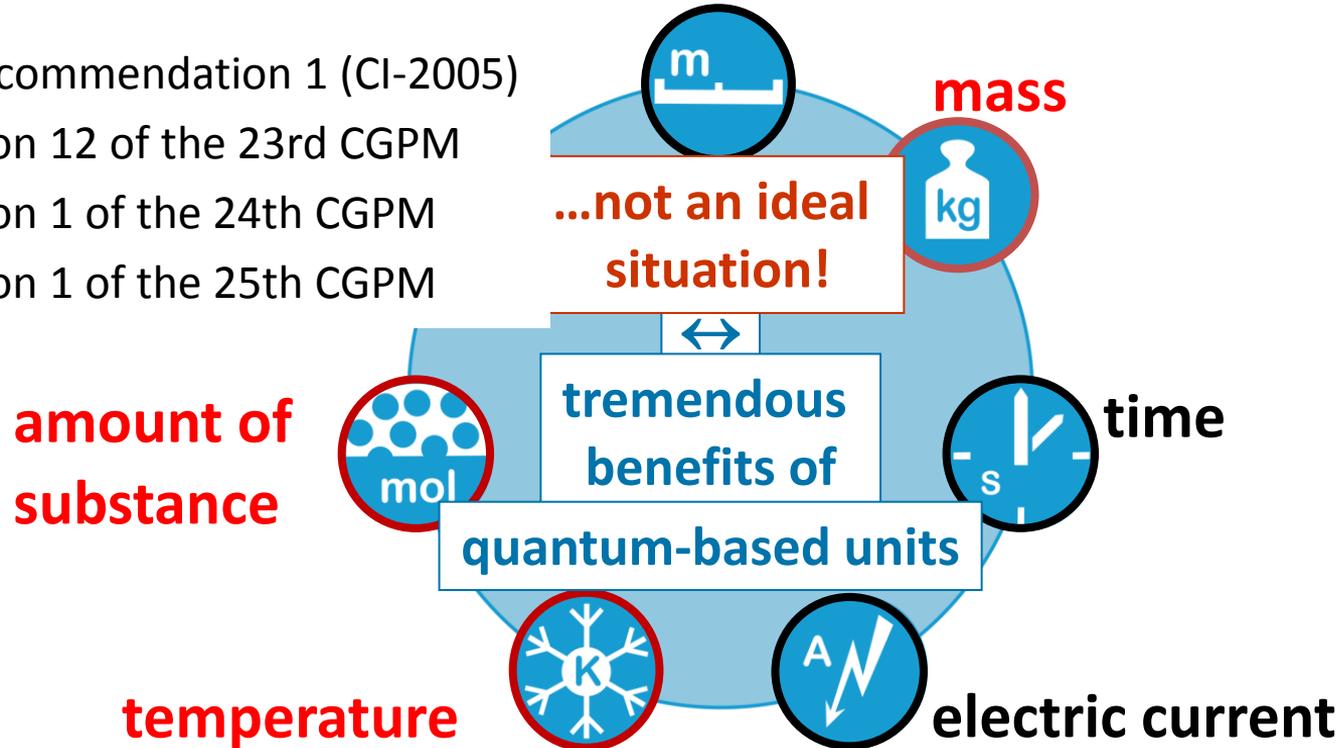
Build a coherent and consistent quantum-based system of units



The International System of Units: SI

Build a coherent and consistent quantum-based system of units

- CIPM Recommendation 1 (CI-2005)
- Resolution 12 of the 23rd CGPM
- Resolution 1 of the 24th CGPM
- Resolution 1 of the 25th CGPM



The International System of Units: SI

Build a coherent and consistent quantum-based system of units

- CIPM Recommendation 1 (CI-2005)
- Resolution 12 of the 23rd CGPM
- Resolution 1 of the 24th CGPM
- Resolution 1 of the 25th CGPM

→ Resolution 1 of the 26th CGPM

→ 13 years and 7 CCU meetings!

Publications

Redefinition of the kilogram: a decision whose time has come, Metrologia 42, 71 (2005).

and

Redefinition of the kilogram, ampere, kelvin and mole: a proposed approach to implementing CIPM recommendation 1 (CI-2005), Metrologia 43, 227 (2006).

Ian M Mills

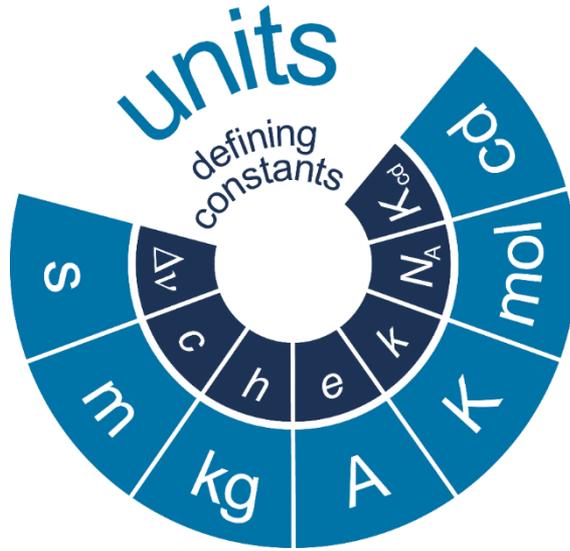
Peter J Mohr

Terry Quinn

Barry N Taylor

Edwin M Williams

Defining Constants



Hyperfine transition ^{133}Cs	$\Delta\nu$
Speed of Light	c
Planck constant	h
Elementary charge	e
Boltzmann constant	k
Avogadro constant	N_A
Luminous Efficacy	K_{cd}



In the 'old' SI:

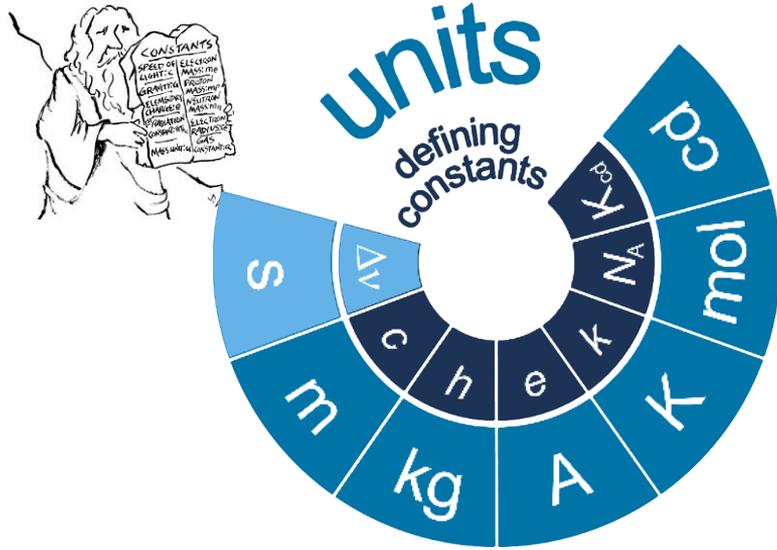
Define the units → Measure the constants (with uncertainty)

...for all times and cultures

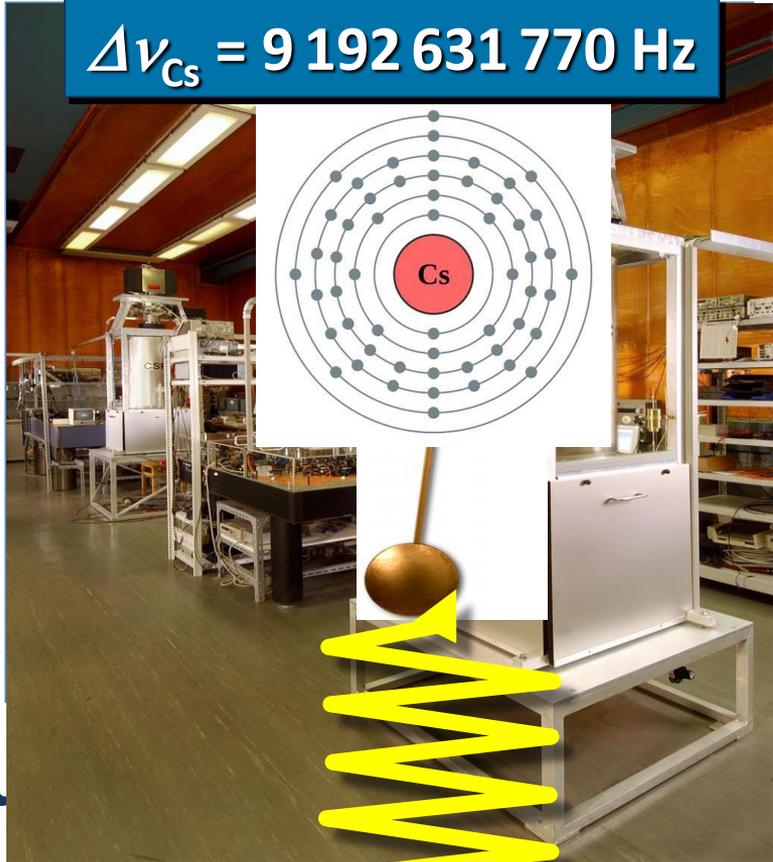
Now in the revised SI:

Fix numerical values of defining constants → Determine the units

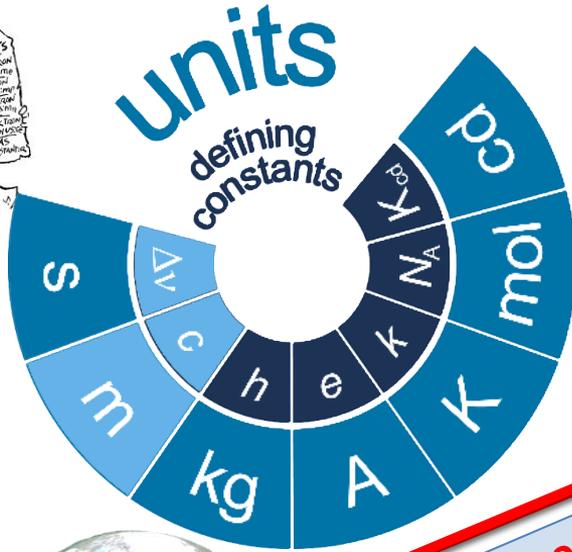
Defining Constants Creating the Units



Relative uncertainty: $2 \cdot 10^{-16}$
(1 second in 160 million years)
...for GPS, science,...

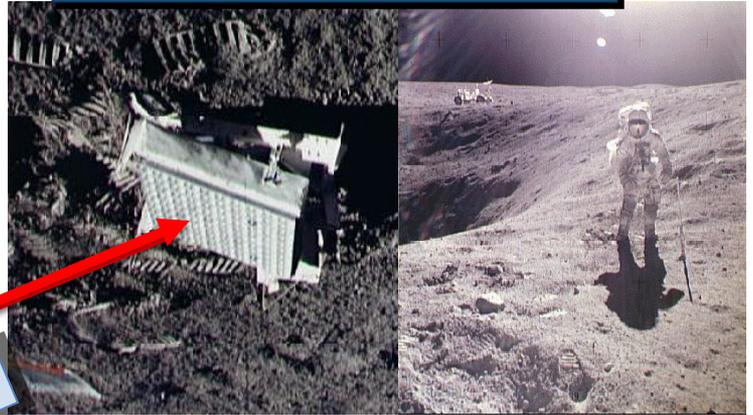


Defining Constants Creating the Units



$d = 384\,402, \dots \text{ km}$

$c = 299\,792\,458 \text{ m/s}$

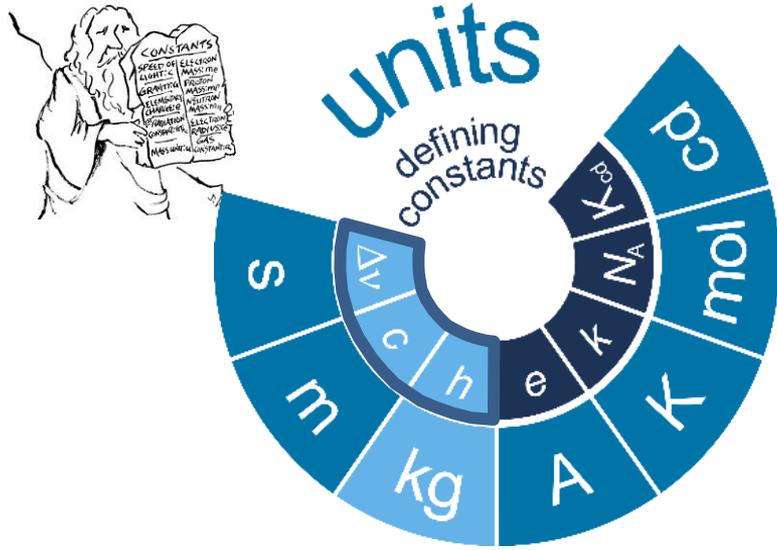


Apollo-11-Crew, 1969

$2t = 2,56 \dots \text{ s} \rightarrow d = c \cdot t$

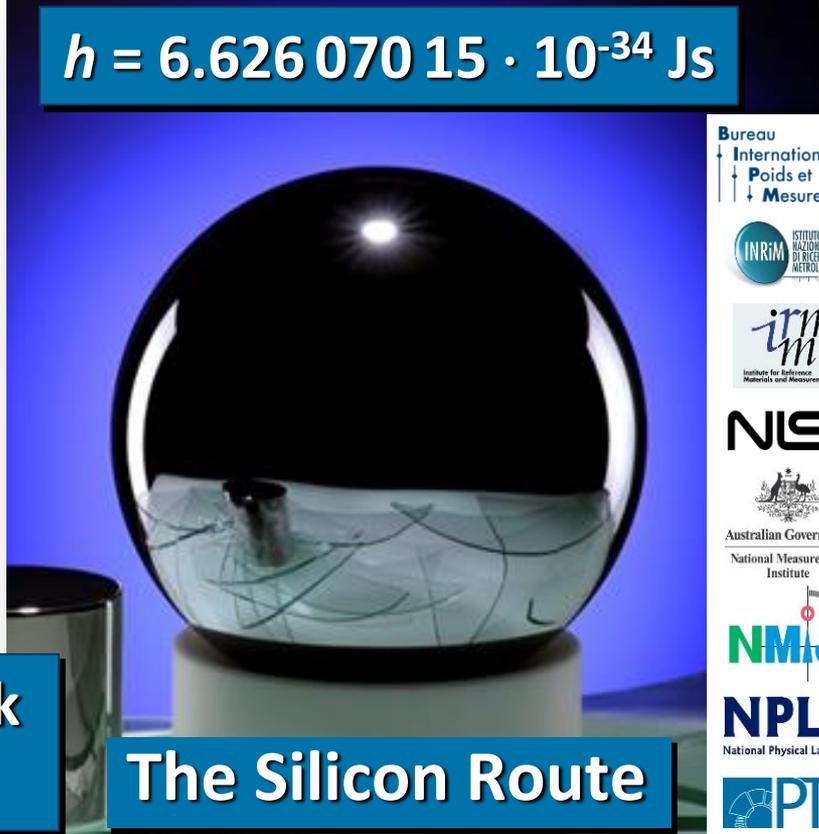


Defining Constants Creating the Units



$$m = n_A \cdot m_e (h, c, \Delta V_{Cs}) \cdot m_A / m_e$$

Quantum mechanics with the Planck constant gives mass to the atoms



The Silicon Route

Bureau International des Poids et Mesures

INRiM Istituto Nazionale di Ricerca Metrologica

irm Institute for Reference Materials and Measurements

NIST

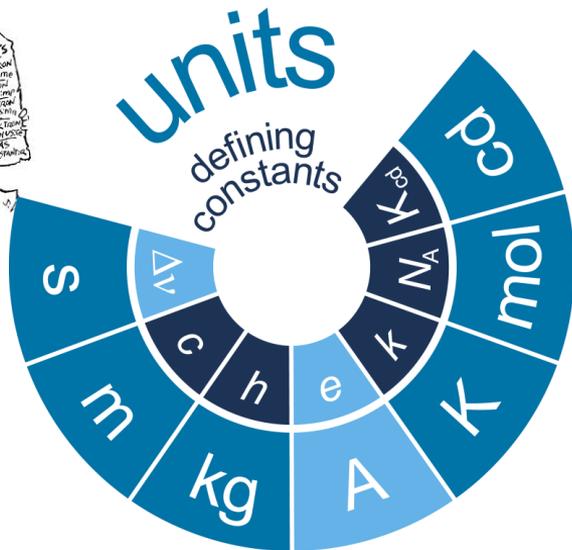
Australian Government National Measurement Institute

NMIJ

NPL National Physical Laboratory

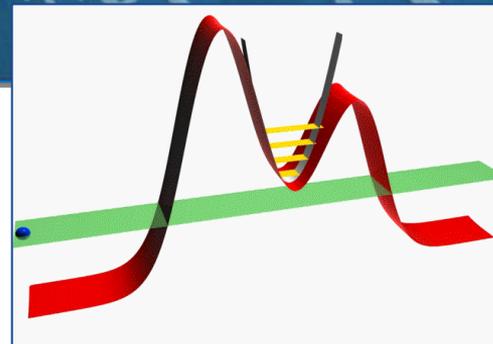
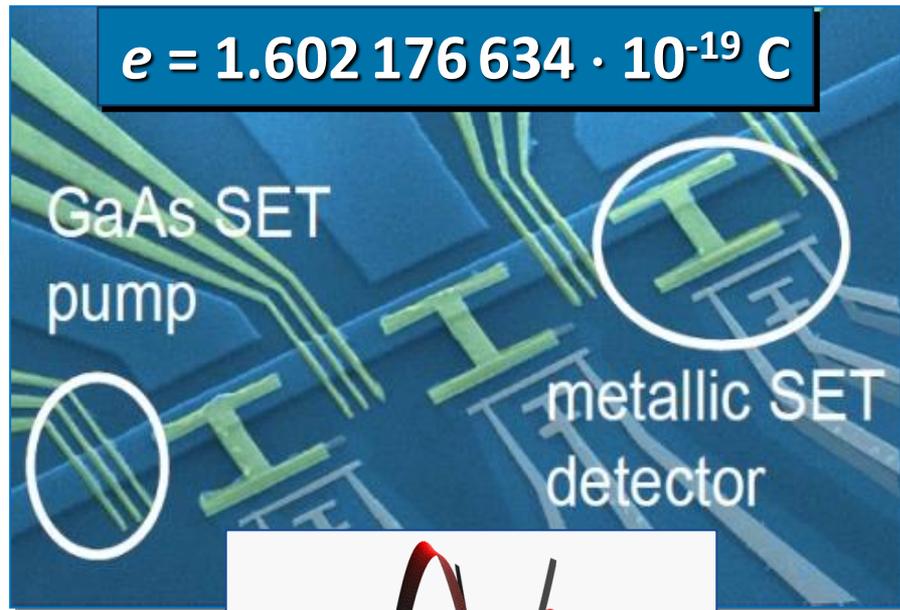
PTB

Defining Constants Creating the Units

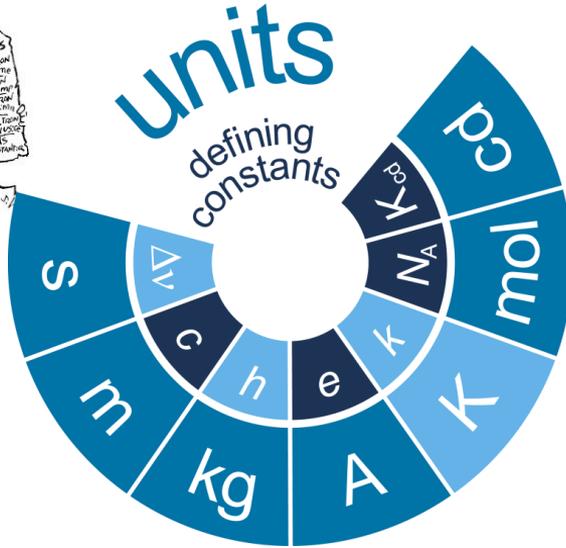


$$I = \langle n \rangle \cdot e \cdot f$$

Count the flow of electrons
in a second



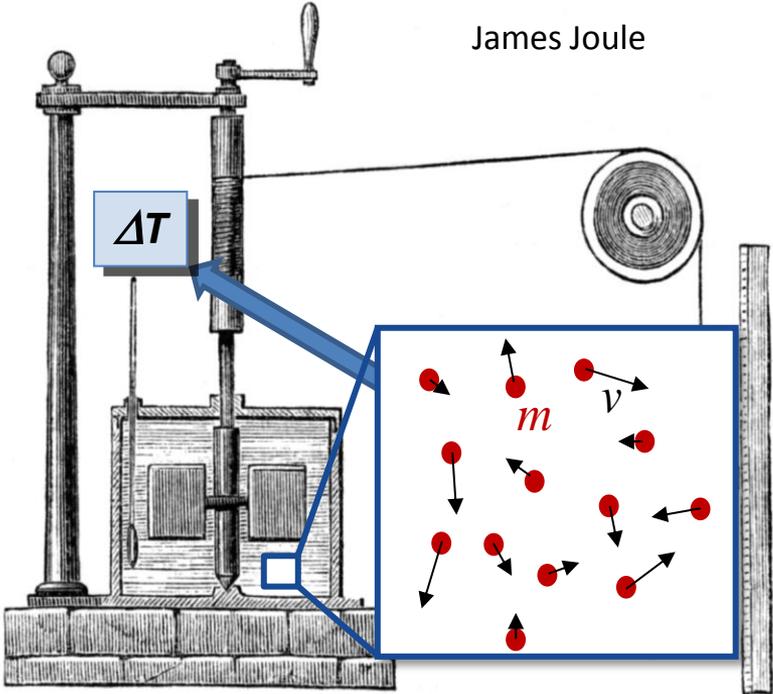
Defining Constants Creating the Units



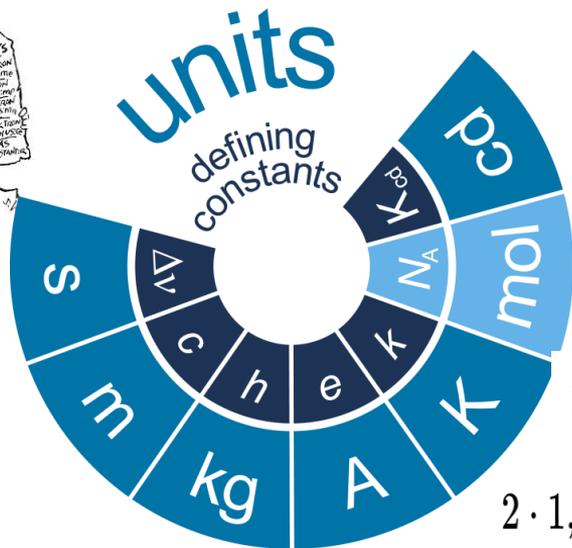
$$\bar{E}_{kin} = (\frac{1}{2})m \cdot \bar{v}^2 = (\frac{3}{2}) \cdot k \cdot T$$

Energy is proportional to Temperature : ideal gas

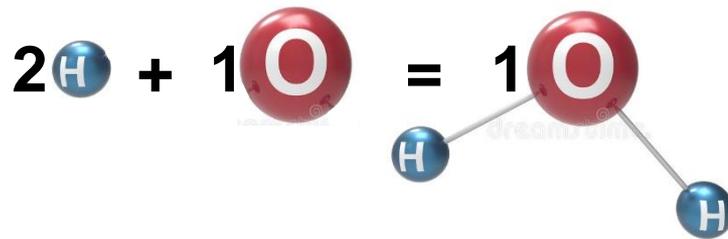
$$k_B = 1.380\,649 \cdot 10^{-23} \text{ J/K}$$



Defining Constants Creating the Units



$$N_A = 6.022\,140\,76 \cdot 10^{23} / \text{mol}$$

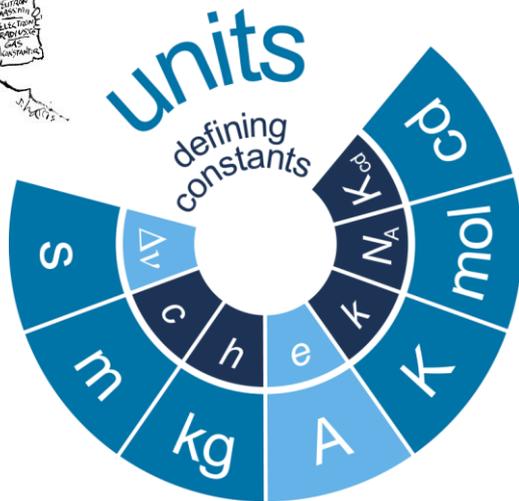


$$2 \cdot 1,00794 \frac{\text{g}}{\text{mol}} + 15,9994 \frac{\text{g}}{\text{mol}} = 18,01528 \frac{\text{g}}{\text{mol}}$$

If you take N_A entities (atoms, molecules) you get 1 mol

Multiply with the atomic/molecular masses: molar mass

Revised International System of Units



$$R_K = \frac{h}{e^2} \approx 25\,813 \, \Omega \Leftrightarrow R_{K-90}$$
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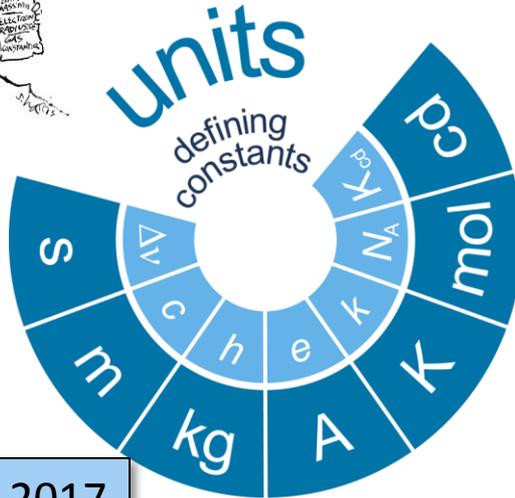
A concept improved fundamentally!

- Guarantees long-time stability
- A set of “defining constants” establish the units in general
- Different realisations
- Realisation everywhere (Universe...)
- Often throughout the entire scale
- Base units are only a convention
- Electric units are “back in the SI”
- Consistency can be ensured
 - Si-kilogram \leftrightarrow Kibble balance
 - Quantum metrological triangle, ...

Revised International System of Units



Establish the constants



CODATA 2017

$$u_{\text{rel}}(e) = 5.2 \times 10^{-9}$$

$$u_{\text{rel}}(k) = 3.7 \times 10^{-7}$$

$$u_{\text{rel}}(h, N_A) = 1.0 \times 10^{-8}$$



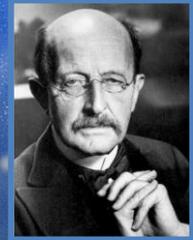
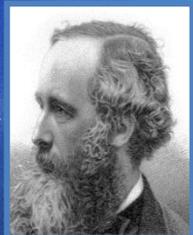
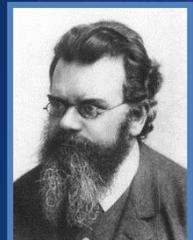
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- Realisation everywhere (Universe...)
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- Base units are only a convention
- Electric units are “back in the SI”
- Consistency can be ensured
- Better experiment → better realization

Ensure continuity, harmonization, stability

SI International System of Units

→ Please see the poster!

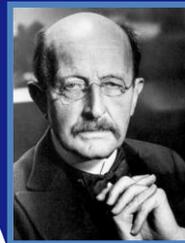
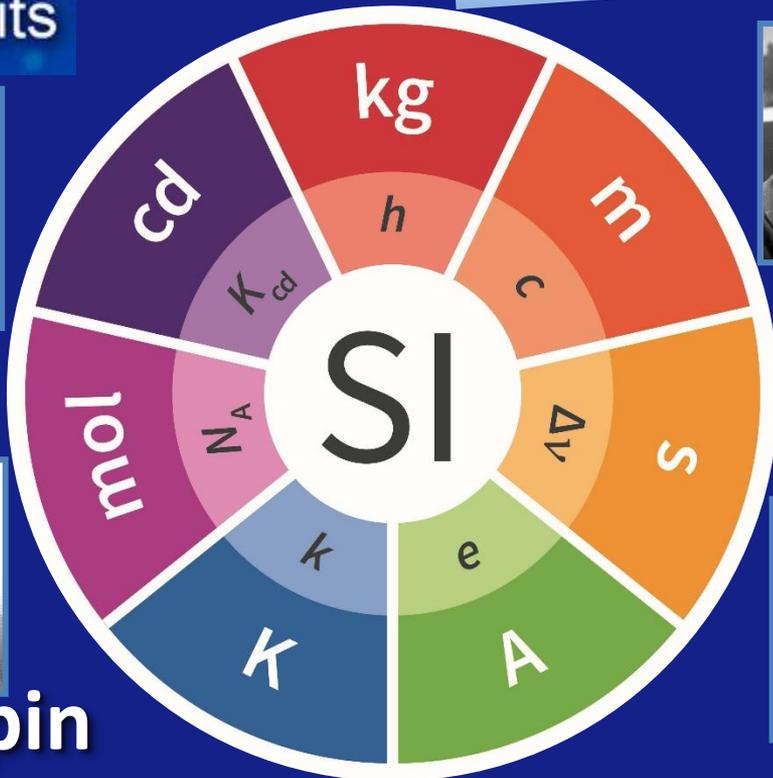
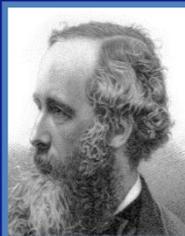
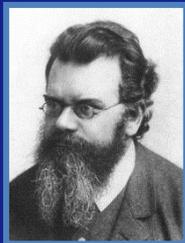


to underpin

...the 2nd Quantum revolution...
From Artefacts to the Quantum SI
...through innovation in science & industry

SI International System of Units

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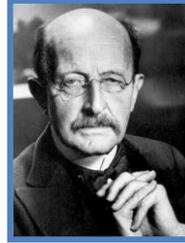
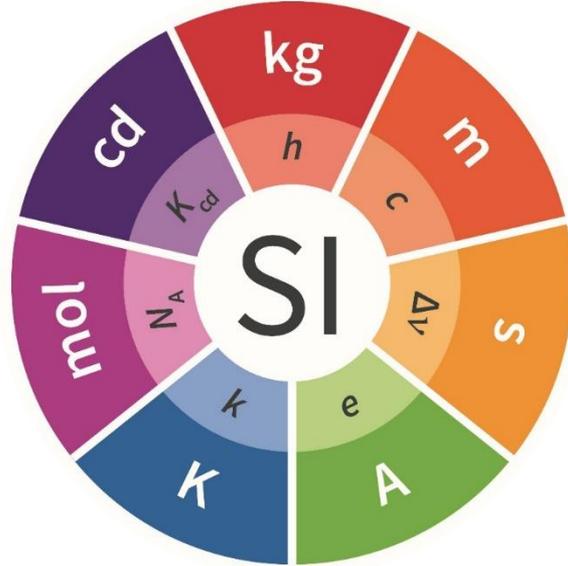
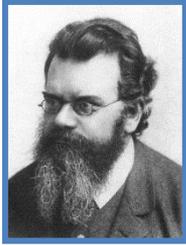


to underpin

...the 2nd Quantum revolution
From Artefacts to the Quantum SI
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SI International System of Units

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to underpin

...the 21st century quantum revolutions...
From Artefacts to the Quantum SI
...through quantum metrology and industry



26^e CGPM

Versailles
13-16 novembre 2018



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Mesures